

Rearrangement of *v*-Triazolo[4,5-*d*]pyrimidines

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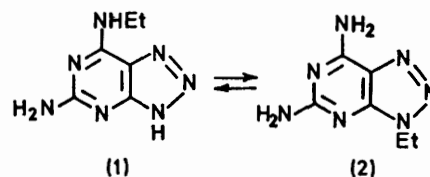
Summary The Dimroth and *retro*-Dimroth rearrangement of a *v*-triazolo[4,5-*d*]pyrimidine system is described.

t.l.c., but, in contrast, refluxing of a solution of (1) in DMAC for *ca.* 18 h gave mainly the rearrangement product

THE Dimroth rearrangement of heterocyclic compounds involves the interchange of exocyclic and endocyclic nitrogen atoms.¹ Conversion of one *v*-triazolo[4,5-*d*]pyrimidine into another is a special type of Dimroth rearrangement involving exocyclic and endocyclic C-N moieties. Previously, the rearrangement of 7-amino-3-benzyl-3*H*-*v*-triazolo[4,5-*d*]pyrimidine to the 7-benzylamino-isomer was unsuccessful.²

The triazolo-pyrimidine (2) was prepared as described by amination of the corresponding 7-chloro-compound.³ Treatment of the known 5-amino-7-chloro-*v*-triazolo[4,5-*d*]pyrimidine⁴ with aqueous ethylamine gave the isomer of (2), (1).

Refluxing of a solution of (2) in dimethylacetamide (DMAC; b.p. 163—165°) for *ca.* 85 h gave unchanged (2) and traces of the rearrangement product (1) identified by



(2) containing only traces of unchanged (1) (t.l.c.).† The rearrangement of (1) was also examined in pyridine (b.p. 112—115°), 2-picoline (b.p. 128°), and 2,4-lutidine (85% minimum; b.p. 155—157°) under reflux to give 65 (18 h), 76 (16 h), and 90% (16 h) conversion into (2), respectively, these values being determined by measurement of the integrated intensities of the MeCH₂ signals for (1) and (2) in the n.m.r. spectra [(CD₃)₂SO; Me₄Si internal reference]

† Traces of unidentified components were observed on many of the chromatograms.

of the products. In pyridine the isomer mixture was not at equilibrium since refluxing of a solution of (1) in this solvent for 132 h gave (2) containing only traces of (1) (t.l.c.). Also, in pyridine the rearrangement appeared to be irreversible because refluxing of a solution of (2) in this solvent for the same period of time (132 h) gave none of (1) (t.l.c.). Presumably this rearrangement involves a 5-diazopyrimidine intermediate, which favours ring closure to the amino-group of greater nucleophilicity.⁵ Refluxing of a solution of (1) in pyridine containing an equivalent amount of β -naphthol to trap the 5-diazopyrimidine intermediate was unsuccessful; the product was a mixture of (1), (2), and β -naphthol (t.l.c.).

The rearrangement (1) \rightarrow (2) depends not only on the temperature, but also on the basicity of the solvent. When a solution of (1) in dibutylamine (b.p. 159–161°; pK_a ca. 11) was heated at 130° for 16 h, no rearrangement to (2) was observed (t.l.c.), which suggested that the anion of (1) undergoes rearrangement less readily than (1) itself. This surmise was confirmed by refluxing a solution of the sodium salt of (1) in 2-picoline for 16 h to give a product that con-

tained no (2) (t.l.c.).[‡] Further, when separate solutions of (1) in pyridine (pK_a^{25} 5.17)⁶ and in 2-picoline (pK_a^{25} 5.97)⁶ were heated at ca. 110° for 18 h, the n.m.r. spectra of the products indicated 37 and 25% conversion into (2), respectively. Presumably, the amount of rearrangement decreases as the concentration of the anion (1) increases. Although this result suggested that the rearrangement of (2) to (1) might be favoured by a strong base, no rearrangement was observed when a solution of (2) was refluxed in dibutylamine for 18 h.⁷ Apparently, however, the system of (1) and (2) in DMAC described above is at equilibrium. Refluxing of a solution of (2) in DMAC containing potassium carbonate for 18 h shifted the equilibrium so that only (1) (potassium salt) was observed (t.l.c.). This result was confirmed by the u.v. spectrum of the crude rearrangement product.

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[‡] The product of this reaction gave analytical data consistent with its formulation as the hemihydrate as the sodium salt of (1).

¹ D. R. Sutherland and G. Tennant, *J. Chem. Soc. (C)*, 1971, 706.

² A. Albert, *J. Chem. Soc. (C)*, 1969, 152.

³ Y. F. Shealy, R. F. Struck, J. D. Clayton, and J. A. Montgomery, *J. Org. Chem.*, 1961, **26**, 4433.

⁴ Y. F. Shealy, J. D. Clayton, C. A. O'Dell, and J. A. Montgomery, *J. Org. Chem.*, 1962, **27**, 4518.

⁵ R. Hull, *J. Chem. Soc.*, 1958, 2746.

⁶ H. C. Brown and X. R. Mihm, *J. Amer. Chem. Soc.*, 1955, **77**, 1723.

⁷ Similarly 1-alkyl-5-amino-1,2,3-triazoles resist Dimroth rearrangement. See ref. 1.